

Highly enantioselective ene-type spiro-cyclization of allyl propargyl ethers catalyzed by cationic palladium(II) complexes with a new type of PN-ligand bearing achiral *gem*-dimethyl oxazoline

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Abstract

The first examples of the asymmetric ene-type spiro-cyclization catalyzed by cationic palladium(II) complexes with a new PN-ligand bearing achiral *gem*-dimethyl oxazoline unit were demonstrated. Spiro-products were synthesized from ether substrates with common, medium and large membered rings with high enantiomeric excesses in almost quantitative yields. © 2002 Elsevier Science B.V. All rights reserved.

Keywords: Asymmetric catalysis; Cyclizations; Enynes; PN-ligands; Palladium; Spiro

1. Introduction

The enantioselective spiro ring construction is an important issue because many natural compounds have chiral spiro centers [1,2]. Some examples of catalytic spiro-cyclizations have been reported by asymmetric intramolecular Mizoroki–Heck reactions [3–7]. In spite of a similar but higher potential, transition metal-catalyzed ene-type carbocyclization has never been applied to asymmetric spiro-cyclizations [8–19].¹ We have already reported that palladium(II)-catalyzed ene-type cyclization proceeds quantitatively

with excellent enantioselectivity generating a new quaternary carbon center [20]. Herein, we wish to report a highly enantioselective spiro ethers formation catalyzed by our cationic chiral palladium(II) complexes with a new PN-ligand bearing achiral oxazoline unit substituted by sterically demanding *gem*-dialkyl groups.

2. Experimental

Typical procedure for palladium(II)-catalyzed spiro-cyclization: thoroughly degassed dimethylsulfoxide (DMSO) (3.0 ml) was injected under argon into a Pyrex Schlenk tube containing [(MeCN)₄Pd](BF₄)₂ (6.6 mg, 0.0150 mmol) and (aS)-PN-ligand **7** (16.0 mg, 0.0300 mmol), and this solution was stirred at room temperature for 5 min. Then **1** (0.300 mmol) and formic acid (5.6 μl, 0.150 mmol) were added, the

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¹ Precedent examples for ene-type spiro-cyclization catalyzed by palladium are limited in achiral version to give only racemic products (see [8]).

tube was sealed with a screw cap. The mixture was stirred at 100 °C. The reaction mixture was washed with brine, and the ether-extracted organic layer was evaporated in vacuo and the residue was purified by short column chromatography (neutral silica-gel, pentane/ether = 10/1) to afford (*S*)-**2** and (*S*)-**3**.

High performance liquid chromatographic (HPLC) analyses were conducted on a JASCO PU-980, LG-980-02, DG-980-50, AS-950 and CO-966 instrument equipped with model UV-975 spectrometers as an ultra violet light; Chiral column were DAICEL-CHIRALCEL AD-H, AS, OB-H, OD-H; peak area were calculated by JASCO-BORWIN (Windows NT) as an automatic integrator. Capillary gas chromatographic (GC) analyses were conducted on a Shimadzu GC-14B instrument equipped with FID detector by using N₂ (75 kPa) as a carrier gas; peak area were calculated by a Shimadzu C-R6A as an automatic integrator; chiral column were CP-Cyclodextrin-β-2,3,6-M-19 (i.d. 0.25 mm × 25 m; CHROMPACK; GL Sciences Inc.) and CP-Chirasil-Dex CB (i.d. 0.32 mm × 25 m; CHROMPACK; GL Sciences Inc.); split ratio was 100:1.

2a: ¹H NMR (300 MHz, CDCl₃) δ 1.93 (m, 1H), 2.11 (m, 1H), 2.40–2.50 (2H), 3.61 (d, *J* = 8.4 Hz, 1H), 3.70 (s, 3H), 3.76 (d, *J* = 9.0 Hz, 1H), 4.74 (dd, *J* = 17.7, 2.4 Hz, 1H), 4.94 (dd, *J* = 17.7, 2.4 Hz, 1H), 5.45 (dt, *J* = 5.4, 2.4 Hz, 1H), 5.63 (t, *J* = 2.7 Hz, 1H), 5.98 (dt, *J* = 6.0, 2.1 Hz, 1H).

3a: ¹H NMR (300 MHz, CDCl₃) δ 2.53 (s, 4H), 3.70 (s, 3H), 3.72 (s, 2H), 4.87 (d, *J* = 2.4 Hz, 2H), 5.68 (bs, 2H), 5.76 (t, *J* = 2.4 Hz, 1H).

2b: ¹H NMR (300 MHz, CDCl₃) δ 1.40–2.22 (6H), 3.50 (d, *J* = 8.4 Hz, 1H), 3.71 (s, 3H), 3.82 (d, *J* = 8.7 Hz, 1H), 4.70 (dd, *J* = 17.7, 2.7 Hz, 1H), 4.98 (dd, *J* = 17.7, 2.7 Hz, 1H), 5.30 (dm, *J* = 10.2 Hz, 1H), 5.67 (t, *J* = 2.4 Hz, 1H), 5.99 (dt, *J* = 10.2, 3.6 Hz, 1H).

3b: ¹H NMR (300 MHz, CDCl₃) δ 1.40–2.22 (6H), 3.63 (d, *J* = 8.7 Hz, 1H), 3.69 (d, *J* = 8.7 Hz, 1H), 3.71 (s, 3H), 4.82 (dd, *J* = 17.7, 2.7 Hz, 1H), 4.90 (dd, *J* = 17.7, 2.7 Hz, 1H), 5.68–5.80 (3H).

2c: ¹H NMR (300 MHz, CDCl₃) δ 1.59–1.85 (6H), 2.00–2.40 (2H), 3.52 (d, *J* = 9.0 Hz, 1H), 3.71 (s, 3H), 3.95 (d, *J* = 8.7 Hz, 1H), 4.72 (dd, *J* = 17.7, 2.7 Hz, 1H), 4.92 (dd, *J* = 17.7, 2.7 Hz, 1H), 5.32 (d, *J* = 12.0 Hz, 1H), 5.72 (t, *J* = 2.7 Hz, 1H), 5.90 (dt, *J* = 11.7, 5.7 Hz, 1H).

3c: ¹H NMR (300 MHz, CDCl₃) δ 1.20–2.40 (8H), 3.64 (d, *J* = 9.0 Hz, 1H), 3.71 (s, 3H), 3.75 (d, *J* = 9.3 Hz, 1H), 4.75–4.91 (2H), 5.67 (m, 1H), 5.70 (t, *J* = 2.7 Hz, 1H), 5.93 (m, 1H).

2d: ¹H NMR (300 MHz, CDCl₃) δ 0.70–2.60 (10H), 3.53 (d, *J* = 8.7 Hz, 1H), 3.70 (s, 3H), 3.98 (d, *J* = 8.7 Hz, 1H), 4.69 (dd, *J* = 17.7, 2.4 Hz, 1H), 4.69 (dd, *J* = 17.7, 2.4 Hz, 1H), 4.91 (d, *J* = 11.7 Hz, 1H), 5.60 (dd, *J* = 12.0, 8.4 Hz, 1H), 5.71 (t, *J* = 2.4 Hz, 1H).

3d: ¹H NMR (300 MHz, CDCl₃) δ 0.80–2.40 (10H), 3.62 (d, *J* = 8.4 Hz, 1H), 3.70 (s, 3H), 3.75 (d, *J* = 8.7 Hz, 1H), 4.79 (dd, *J* = 17.7, 2.4 Hz, 1H), 4.88 (dd, *J* = 17.7, 2.4 Hz, 1H), 5.73 (dt, *J* = 10.8, 8.4 Hz, 1H), 5.68 (t, *J* = 2.7 Hz, 1H), 5.85 (dt, *J* = 10.5, 8.1 Hz, 1H).

2e: ¹H NMR (300 MHz, CDCl₃) δ 1.00–2.20 (24H), 3.67 (d, *J* = 8.4 Hz, 1H), 3.70 (s, 3H), 3.84 (d, *J* = 8.7 Hz, 1H), 4.80 (d, *J* = 2.7 Hz, 2H), 5.33 (d, *J* = 15.9 Hz, 1H), 5.46 (dt, *J* = 15.3, 6.6 Hz, 1H), 5.69 (t, *J* = 2.7 Hz, 1H).

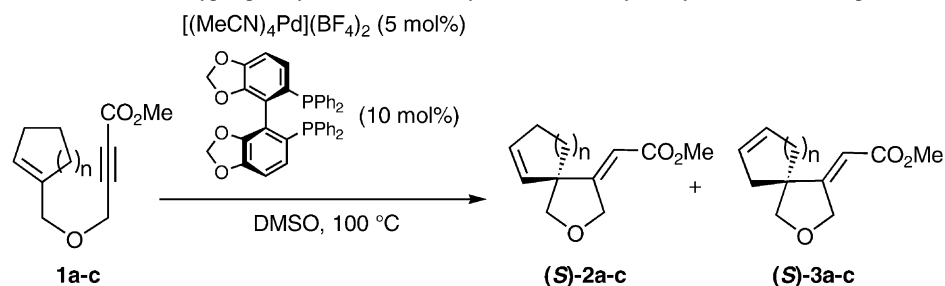
3. Results and discussion

First, in Pd^{II}-catalyzed spiro-cyclizations, we employed bidentate C₂-symmetric PP-ligand such as (*R*)-SEGPHOS ((4,4'-bi-1,3-benzodioxole)-5,5'-diylbis(diphenylphosphine)) [21], that has been found to be so effective to substrates with acyclic olefin, for a variety of cyclic allyl propargyl ethers **1** under polar conditions ([MeCN]₄Pd(BF₄)₂/DMSO) (Table 1). However, reactions were rather slow, taking 72 h for full conversion. Enantioselectivities were unexpectedly low (21–38% e.e.), although the common membered spiro-products were obtained in good yields. Moreover, the olefin-migrations of the primary generated product **2** occurred in all cases leading to the secondary olefin regioisomer **3**.

Based on our mechanistic studies about the transition states for these C–C bond formation, we have already devised bidentate C₁-symmetric PN-ligands having binaphthyl backbone [22].² Therefore, PN-ligands bearing a chiral oxazoline unit [23–28] were investigated for substrate **1c** in Pd^{II}-catalyzed

² PN-ligands have been known to have advantages to prevent olefin-migrations in asymmetric Mizoroki–Heck reactions [29,30].

Table 1

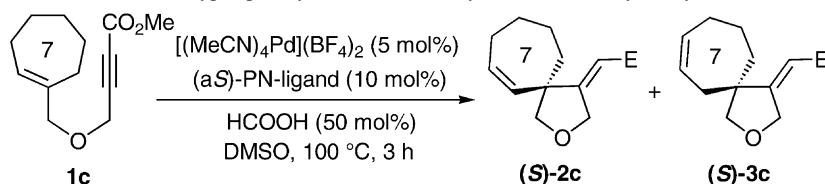
Enantioselective ene-type spiro-cyclization of 1,6-enyne ethers **1** catalyzed by cationic Pd^{II} complex with (*R*)-SEGPPOS^a

Entry	Substrate	Reaction time (h)	Yield (%) (e.e. (%) ^b)		Total e.e. value (%)
			2	3	
1	1a (<i>n</i> = 1)	72	35 (23)	21	23
2	1b (<i>n</i> = 2)	72	20 (8)	52 (27)	21
3	1c (<i>n</i> = 3)	16	0 (–)	99 (38)	38

^a Reactions were carried out in thoroughly-degassed DMSO at 100 °C with 5 mol% of [(MeCN)₄Pd](BF₄)₂ and 10 mol% of (*R*)-SEGPPOS.

^b The e.e. value were based on chiral GC and/or HPLC analyses.

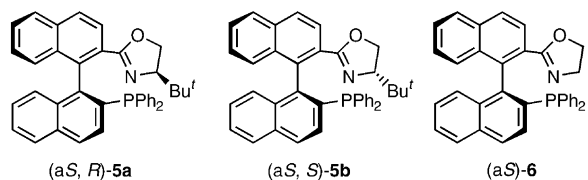
Table 2

Enantioselective ene-type spiro-cyclization of 1,6-enyne ether **1c** catalyzed by cationic Pd^{II} complexes with chiral PN-ligand^a

Entry	PN-ligand	Yield (%) (e.e. (%) ^b)		Total e.e. value (%)
		2c	3c	
1	(<i>aS</i> , <i>R</i>)- 5a	22 (96)	77 (78)	82
2	(<i>aS</i> , <i>S</i>)- 5b	32 (88)	65 (76)	80
3	(<i>aS</i>)- 6	58 (61)	23 (27)	51

^a Reaction were carried out in thoroughly-degassed DMSO at 100 °C with 5 mol% of [(MeCN)₄Pd](BF₄)₂, 10 mol% of PN-ligand and 50 mol% of HCOOH.

^b The e.e. value were based on chiral GC and/or HPLC analyses.



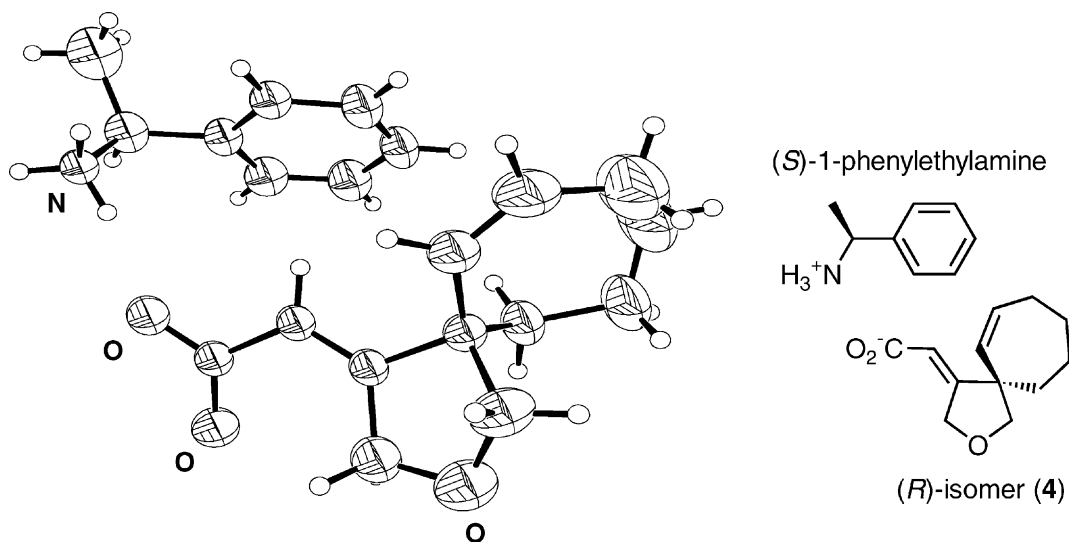


Fig. 1. ORTEP representation and structural formula of the (*S*)-1-phenylethylamine salt ((*R*)-derivative (**4**) from (*aR*)-PN-ligands).

spiro-cyclizations. In sharp contrast to PP-ligand, PN-ligands **5a** and **5b** having a *tert*-Bu oxazoline unit were found to be effective to give the spiro-product **2c** with (*S*)-sense in excellent yields and good enantioselectivities (over 80% e.e.) with olefin-migrations (Table 2, entries 1 and 2). The absolute configuration of **2c** was determined by X-ray crystallographic analysis,³ as a chiral amine salt of the corresponding carboxylic acid **4** with (*S*)-1-phenylethylamine. ORTEP drawings are shown in Fig. 1. On the other hand, PN-ligand **6** having no

³ **2c** was converted to the corresponding carboxylic acid **4** through hydrolysis, and then crystallized as a diastereomeric salt of (*S*)-1-phenylethylamine in a CH₂Cl₂–hexane–AcOEt mixture at room temperature. Crystal data for this salt in X-ray analysis: formula C₁₂H₁₆O₃·C₈H₁₁N, orthorhombic, space group *P*2₁2₁2₁ (#19), *a* = 14.205(2) Å, *b* = 19.760(3) Å, *c* = 6.735(3) Å, *V* = 1890.5(10) Å³, *Z* = 4, and *D* = 1.157 g/cm³. X-ray diffraction data were collected on a Rigaku AFC7R diffractometer with graphite-monochromated Mo Kα (*λ* = 0.71069 Å) at –20 °C and the structure was solved by direct methods (SIR97). The final cycle of full-matrix least-squares refinement was based on 2492 observed reflections (*I* > 3σ(*I*)) and 226 variable parameters and converged to *R* = 0.0447 and *R*_w = 0.1358. Crystallographic data (excluding structure factors) for the structure reported in this paper have been deposited with the Cambridge Crystallographic Data Center as supplementary publication no. CCDC-188150. Copies of the data can be obtained free of charge on application to CCDC, 12 Union Road, Cambridge CB21EZ, UK (fax: +44-1223-336-033; e-mail: deposit@ccdc.cam.ac.uk).

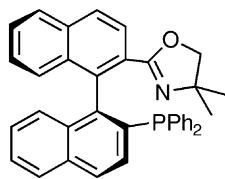
alkyl substituent in oxazoline unit showed poor enantiomeric excess (total 51% e.e., entry 3). These results imply that the center of chirality (*R* or *S*) at the 4-position of the oxazoline is *not* important but the presence of sterically demanding alkyl substituent is necessary to achieve higher enantioselectivity.

Finally, our effective (*aS*)-PN-ligand **7**, doubly substituted by methyl groups, was executed. For 7-membered ring **1c**, PN-ligand **7** provided the corresponding spiro-products **2c** and **3c** with higher enantiomeric excess (total 84% e.e.) and in excellent yield (Table 3, entry 3). Even for other membered rings, spiro-cyclization proceeded successfully; as a total value, 88% e.e. and 95% yield for 5-membered ring **1a** (entry 1), 71% e.e. and 83% yield for 6-membered ring **1b** (entry 2). For **1a** with PN-ligand **7**, olefin-migration was prevented, so that enantio-enriched **2a** was major product with isomeric **3a**. Medium 8-membered ring substrate **1d** also cyclized successfully in 84% e.e. and 94% yield (entry 4). For large ring system of 15-membered ring **1e**, spiro-cyclization proceeded successfully not accompanying with the olefin-migration, to afford the single product **2e** in 83% e.e. and in almost quantitative yield (entry 5).⁴

⁴ In the presence of 50 mol% of HCOOH, side-product which was reduced in acetylene unit to olefin was obtained in accompany with the desired product **2e**. The presence of HCOOH retards the olefin migrations in all cases.

Table 3

Enantioselective ene-type spiro-cyclization of 1,6-enyne ethers **1** catalyzed by cationic Pd^{II} complexes with (aS)-gem-dimethyl PN-ligand **7**^a



(aS)-7

Entry	Substrate (ring size)	Reaction time (h)	Yield (%) (e.e. (%) ^b)		Total e.e. value (%)
			2	3	
1	1b (5)	3	88 (88)	7	88
2	1b (6)	22	63 (84)	20 (31)	71
3	1c (7)	3	13 (88)	78 (83)	84
4	1d (8)	6	33 (93)	61 (80)	84
5 ^c	1e (15)	11	>90 (83)	0 (–)	83

^a Reactions were carried out in thoroughly-degassed DMSO at 100 °C with 5 mol% of [(MeCN)₄Pd](BF₄)₂, 10 mol% of PN-ligand **7** and 50 mol% of HCOOH unless otherwise noted.

^b The e.e. value were based on chiral GC and/or HPLC analyses.

^c In the absence of HCOOH.

4. Conclusions

We have established highly efficient ene-type spiro-cyclization of allyl propargyl ethers catalyzed by cationic palladium(II) with a new PN-ligand bearing achiral gem-dimethyl oxazoline unit. Wide scope of spiro-compounds, from common to large membered ether rings, were synthesized in high to excellent enantioselectivities almost quantitatively. This is the first example of the asymmetric ene-type spiro-cyclization and of great potential for other ring formations including heterocycles and carbocycles.

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References

- [1] S.F. Martin, *Tetrahedron* 36 (1980) 419–460.
- [2] K. Fuji, *Chem. Rev.* 93 (1993) 2037–2066.
- [3] A. Ashimori, L.E. Overman, *J. Org. Chem.* 57 (1992) 4571–4572.
- [4] L.E. Overman, D.J. Poon, *Angew. Chem. Int. Ed. Engl.* 36 (1997) 518–521.
- [5] L. Ripa, A. Hallberg, *J. Org. Chem.* 62 (1997) 595–602.
- [6] A. Ashimori, B. Bachand, L.E. Overman, D.J. Poon, *J. Am. Chem. Soc.* 120 (1998) 6477–6487.
- [7] E. Negishi, C. Coperet, S. Ma, S.Y. Liou, F. Liu, *Chem. Rev.* 96 (1996) 365–393.
- [8] B.M. Trost, M. Lautens, *J. Am. Chem. Soc.* 107 (1985) 1781–1783.
- [9] B.M. Trost, M.J. Krische, *Synlett* (1998) 1–16.
- [10] B.M. Trost, *Acc. Chem. Res.* 23 (1990) 34–42.
- [11] B.M. Trost, *Chem. Eur. J.* 4 (1998) 2405–2412.
- [12] C. Aubert, O. Buisine, M. Malacria, *Chem. Rev.* 102 (2002) 813–834.
- [13] B.M. Trost, D.C. Lee, F. Rise, *Tetrahedron Lett.* 30 (1989) 651–654.
- [14] B.M. Trost, B.A. Czeskis, *Tetrahedron Lett.* 35 (1994) 211–214.
- [15] A. Goeke, M. Sawamura, R. Kuwano, Y. Ito, *Angew. Chem. Int. Ed. Engl.* 35 (1996) 662–663.
- [16] P. Cao, X. Zhang, *Angew. Chem. Int. Ed. Engl.* 39 (2000) 4104–4106.
- [17] A. Lei, M. He, X. Zhang, *J. Am. Chem. Soc.* 124 (2002) 8198–8199.
- [18] Q. Zhang, X. Lu, *J. Am. Chem. Soc.* 122 (2000) 7604–7605.
- [19] Q. Zhang, X. Lu, X. Han, *J. Org. Chem.* 66 (2001) 7676–7684.
- [20] M. Hatano, M. Terada, K. Mikami, *Angew. Chem. Int. Ed. Engl.* 40 (2001) 249–253.
- [21] T. Saito, T. Yokozawa, T. Ishizaki, T. Moroi, N. Sayo, T. Miura, H. Kumibayashi, *Adv. Synth. Catal.* 343 (2001) 264–267; EP 850945A (1998), US 5872273 (1999).
- [22] M. Hatano, K. Mikami, *Angew. Chem. Int. Ed. Engl.*, submitted for publication.
- [23] M. Ogasawara, K. Yoshida, H. Kamei, K. Kato, Y. Uozumi, T. Hayashi, *Tetrahedron Asymm.* 9 (1998) 1779–1787.
- [24] M. Ogasawara, K. Yoshida, T. Hayashi, *Heterocycles* 52 (2000) 195–201.
- [25] Y. Imai, W. Zhang, T. Kida, Y. Nakatsuji, I. Ikeda, *Tetrahedron Lett.* 39 (1998) 4343–4346.
- [26] K. Selvakumar, M. Valentini, M. Wörle, P.S. Pregosin, A. Albinati, *Organometallics* 18 (1999) 1207–1215.
- [27] K. Selvakumar, M. Valentini, P.S. Pregosin, A. Albinati, F. Eisenräger, *Organometallics* 19 (2000) 1299–1307.
- [28] P. Dotta, A. Magistrate, U. Rothlisberger, P.S. Pregosin, A. Albinati, *Organometallics* 21 (2002) 3033–3041.
- [29] O. Loiseleur, P. Meier, A. Pfaltz, *Angew. Chem.* 108 (1996) 218–220.
- [30] O. Loiseleur, P. Meier, A. Pfaltz, *Angew. Chem. Int. Ed. Engl.* 35 (1996) 200–201.